

A Novel Method for the Synthesis of Anomerically Allylated C-Glycofuranosides. An Unusual Lewis Acid-catalysed Rearrangement of an Oxetanosyl-*N*-glycoside† to Furanosyl-C-glycosides

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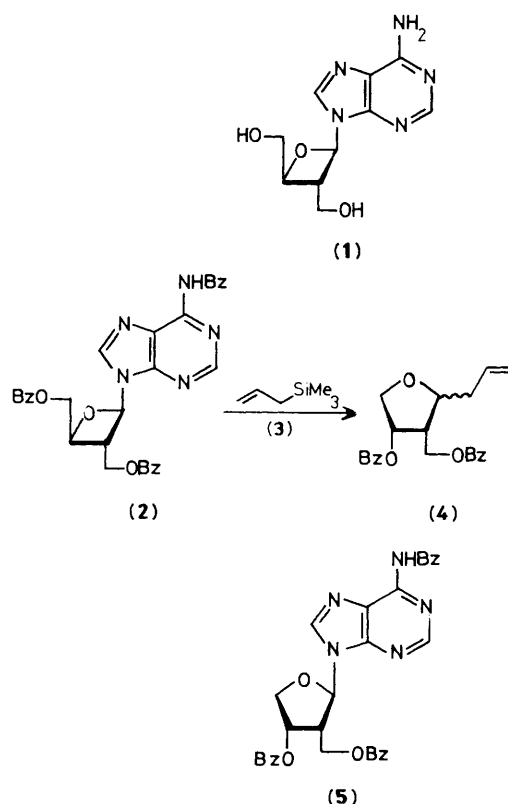
An oxetanosyl-*N*-glycoside is converted into furanosyl-*C*-allylglycosides with allyltrimethylsilane in the presence of Lewis acid.

Many reactions of allylic trimethylsilanes involve electrophilic substitution with allylic rearrangement and loss of the trimethylsilyl group to make a carbon-carbon bond. Recently these bond formation reactions have been applied to the synthesis of *C*-glycosides, particularly *C*-allylglycosides,¹ owing to the potential use of such materials in the total synthesis of natural products such as the pseudomonic acids² and palytoxin.³ Further, epoxides⁴ and oxetanes⁵ react with allylic trimethylsilanes in the presence of Lewis acids to give homoallylic alcohols and hex-5-en-1-ols, respectively.

In continuation of our studies on the chemical properties of the oxetane ring of oxetanocin (**1**),⁶ we have found that the oxetanosyl-*N*-glycoside was converted into furanosyl-*C*-allylglycosides with allyltrimethylsilane.

When *N*-benzoyloxetanocin dibenzoate (**2**) was treated with allyltrimethylsilane (**3**) in the presence of SnCl₄ or BF₃-OEt₂ in MeCN, 2-deoxy-2-hydroxymethyl-*D*-erythrofuranosyl-1-allylglycoside dibenzoates, α - and β -(**4**),[‡] were obtained in

moderate yields (Scheme 1). In the absence of (**3**), however, *N*⁶-benzoyl-2-deoxy-2-hydroxymethyl- β -*D*-erythrofuranosyl-adenine dibenzoate (**5**)[‡] was isolated as the sole product (Table 1). As expected, (**5**) could not be converted into (**4**)



Scheme 1. Bz = PhCO.

[†] Oxetanosyl is taken to refer to the four-membered sugar ring, analogous to furanosyl.

[‡] ¹H N.m.r. data (400 MHz, CDCl₃) for α -(**4**): δ 2.42–2.50 (2H, complex), 2.82 (1H, m), 3.91 (1H, dd, *J* 2.93, 10.74 Hz), 4.25–4.34 (2H, complex), 4.43 (1H, dd, *J* 5.86, 10.74 Hz), 4.61 (1H, dd, *J* 5.86, 11.23 Hz), 5.14 (1H, dd, *J* 1.95, 10.26 Hz), 5.21 (1H, dd, *J* 1.47, 17.09 Hz), 5.58 (1H, m), 5.91 (1H, m), 7.43–7.47 (4H, complex), 7.55–7.61 (2H, complex), and 8.03–8.09 (4H, complex). β -(**4**): δ 2.53–2.61 (3H, complex), 3.91 (1H, dd, *J* 6.84, 12.69 Hz), 4.07 (1H, dd, *J* 4.88, 10.74 Hz), 4.15 (1H, dd, *J* 1.46, 10.74 Hz), 4.48 (1H, dd, *J* 5.86, 11.23 Hz), 4.54 (1H, dd, *J* 6.35, 11.23 Hz), 5.12–5.21 (2H, complex), 5.45–5.47 (1H, m), 5.88–5.94 (1H, m), 7.41–7.48 (4H, complex), 7.55–7.61 (2H, complex), and 8.02–8.07 (4H, complex). (**5**): δ 3.64–3.74 (1H, m), 4.51 (1H, dd, *J* 2.89, 10.80 Hz), 4.58 (1H, dd, *J* 4.36, 10.80 Hz), 4.69 (1H, dd, *J* 6.12, 11.49 Hz), 4.80 (1H, dd, *J* 5.70, 11.49 Hz), 5.64 (1H, ddd, *J* 2.53, 2.89, 4.36 Hz), 6.47 (1H, d, *J* 3.26 Hz), 7.37–7.66 (9H, complex), 7.85–7.92 (2H, complex), 7.96–8.09 (4H, complex), 8.34 (1H, s), 8.69 (1H, s), and 9.07 (1H, br.s).

Table 1. Lewis acid catalysed rearrangement of (2).

Entry	(2) (mmol)	(3) (mmol)	Acid (mmol)	Products ^a	Ratio α/β^b	Total yield (%) ^c
1	1	10	SnCl ₄ (5) ^d	(4)	1/5	43
2	1	10	BF ₃ OEt ₂ (5) ^d	(4)	1/6	52
3	1	0	SnCl ₄ (2) ^e	(5)	—	16
4	1	0	BF ₃ -OEt ₂ (2) ^f	(5)	—	48

^a All new compounds α - and β -(4), and (5) exhibited satisfactory analytical and spectral data. ^b Anomeric configuration was confirmed by ¹H n.m.r. data given by COSY, long-range COSY, and NOESY (400 MHz). ^c Yield of chromatographically purified products. ^d 30 min at 0 °C. ^e 85 min at room temp. ^f 120 min at room temp.

under the conditions used in entries 1 and 2 of Table 1. These results show that the oxetane ring in (1) was easily cleaved by a Lewis acid to give furanosyl compounds by a ring expansion accompanying transglycosidation. All attempts to react (2) with other silyl nucleophiles such as Me₃SiN₃, Me₃SiCN, and ketene silylacetals were unsuccessful and only (5) was obtained.

In conclusion, this is to the best of our knowledge the first example of the preparation of C-allylglycosides from an N-glycoside. §

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§ Similar results were obtained with allyltributyltin. The reaction mechanism has not been clarified yet.

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